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Key indicators

Single-crystal X-ray study
T = 297 K
 Mean $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$
R factor = 0.036
wR factor = 0.091
 Data-to-parameter ratio = 12.9

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

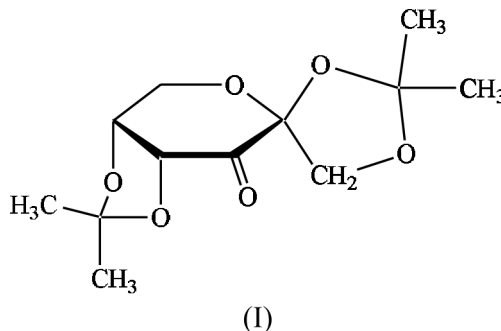
1,2:4,5-Di-*O*-isopropylidene- β -*D*-erythro-hexo-2,3-diulo-2,6-pyranose

The crystal structure of title compound, $\text{C}_{12}\text{H}_{18}\text{O}_6$, has been determined using single-crystal data. There are two independent molecules in the asymmetric unit. NMR coupling constants, dihedral angles as well as puckering parameters indicate the 3S_0 conformation of the pyranose ring in both molecules *A* and *B*. The *cis*-fused 1,3-dioxolane ring in both molecules *A* and *B*, adopts an E_4 conformation. The second five-membered 1,3-dioxolane ring, in molecule *A*, has an E_2 conformation, whereas in molecule *B*, this 1,3-dioxolane ring adopts an E_4 conformation. Molecules *A* form a zigzag infinite chain along the *b* axis with molecules *B* as side chains. This explains why the displacement ellipsoids of atoms of molecule *B* are substantially larger than those of molecule *A*.

Received 8 June 2001
 Accepted 27 June 2001
 Online 6 July 2001

Comment

As a continuation of our research on the synthesis and structure determination of selected sugar amino derivatives (Koš, Steiner, Langer *et al.*, 2000; Koš, Steiner, Gajdoš *et al.*, 2000; Steiner *et al.*, 1998), we have focused attention on 1,2:4,5-di-*O*-isopropylidene- β -*D*-erythro-hexo-2,3-diulo-2,6-pyranose, (I), as a potential starting material. This compound was prepared in two steps from *D*-fructose according to the procedure of Wang *et al.* (1997). The atom-numbering schemes of both molecules are presented in Fig. 1. The conformation of the pyranose ring of (I) is mentioned only twice in the literature. One proposal, based on a molecular modeling study, favored the 0S_3 conformation (McDonald, 1967), whereas a second possibility (more probable), based on the inspection of dihedral angles (from a molecular model) and coupling constants (from NMR), was the 3S_0 conformation (Hervé du Penhoat & Perlin, 1979).



Correlation of dihedral angles associated with coupling constants $J_{5,6} = 2.2 \text{ Hz}$ and $J_{5,6'} = 0.9 \text{ Hz}$ with corresponding dihedral angles ($\text{H}5\text{A}-\text{C}5\text{A}-\text{C}6\text{A}-\text{H}6\text{A}1$ and $\text{H}5\text{B}-\text{C}5\text{B}-\text{C}6\text{B}-\text{H}6\text{B}1$ about -70° , and $\text{H}5\text{A}-\text{C}5\text{A}-\text{C}6\text{A}-\text{H}6\text{A}2$ and $\text{H}5\text{B}-\text{C}5\text{B}-\text{C}6\text{B}-\text{H}6\text{B}2$ about 46°) obtained from X-ray

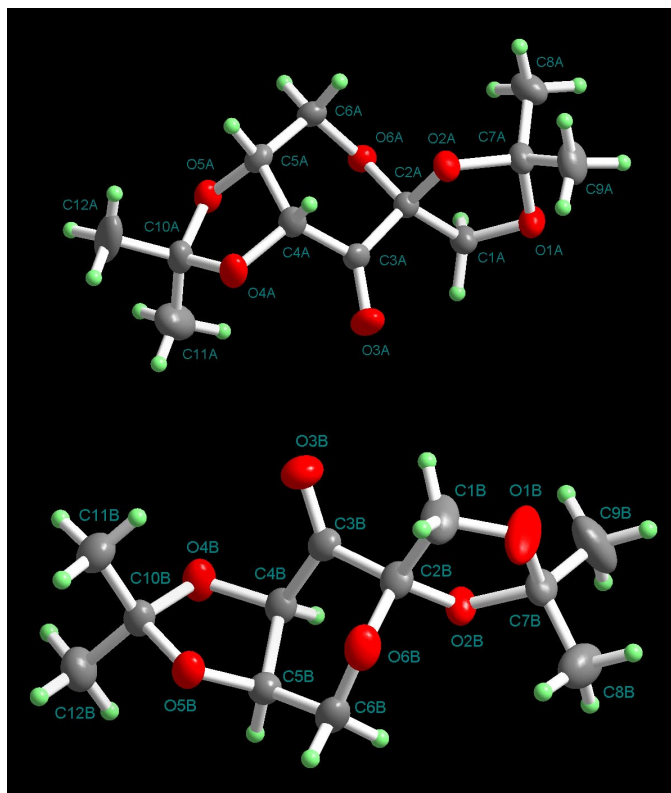


Figure 1
The atom-numbering scheme and displacement ellipsoids at the 30% probability level for molecules A and B of (I).

crystallography as well as puckering parameters [$Q = 0.476(2) \text{ \AA}$, $\varphi = 193.5(8)^\circ$, $\theta = 162.0(2)^\circ$ for molecule A, and $Q = 0.511(2) \text{ \AA}$, $\varphi = 218.2(8)^\circ$, $\theta = 162.8(2)^\circ$ for molecule B] clearly indicate the 3S_0 conformation of the pyranose ring (O6–C2–C3–C4–C5–C6) in both molecules A and B (Fig. 1). The *cis*-fused 1,3-dioxolane ring (C4–O4–C10–O5–C5 in both molecules A and B) adopts the E_4 conformation, where the C4, O4, C10 and O5 atoms lie almost in a plane [for molecule A, dihedral angle C4–O4–C10–O5 = $-3.2(2)^\circ$, and puckering parameters $Q = 0.347(2) \text{ \AA}$ and $\varphi = 247.7(4)^\circ$; for molecule B, dihedral angle C4–O4–C10–O5 = $-2.9(2)^\circ$, and puckering parameters $Q = 0.367(2) \text{ \AA}$ and $\varphi = 248.4(3)^\circ$] and C5 is directed below this plane. The second five-membered 1,3-dioxolane ring (O1–C1–C2–O2–C7 in the molecule A) has an E_2 conformation, where the O1, C1, C2, and O2 atoms lie almost in a plane [dihedral angle O1–C1–C2–O2 is $3.4(2)^\circ$, and puckering parameters $Q = 0.291(2) \text{ \AA}$ and $\varphi = 330.1(4)^\circ$] and C7 faces below this plane. However, in molecule B, this 1,3-dioxolane ring adopts the E_4 conformation, where the O2, C7, O1, and C1 atoms lie in a plane [dihedral angle O2–C7–O1–C1 is $0.8(3)^\circ$, and puckering parameters $Q = 0.171(2) \text{ \AA}$ and $\varphi = 255.7(9)^\circ$] with C2 directed below this plane. It is also noteworthy, that both molecules in the asymmetric unit have the same absolute configuration: *S* on C2, and *R* on C4 and C5.

There are also C6A–H6A1 \cdots O2B and C8A–H8A2 \cdots O1A hydrogen bonds (Table 2) which stabilize the

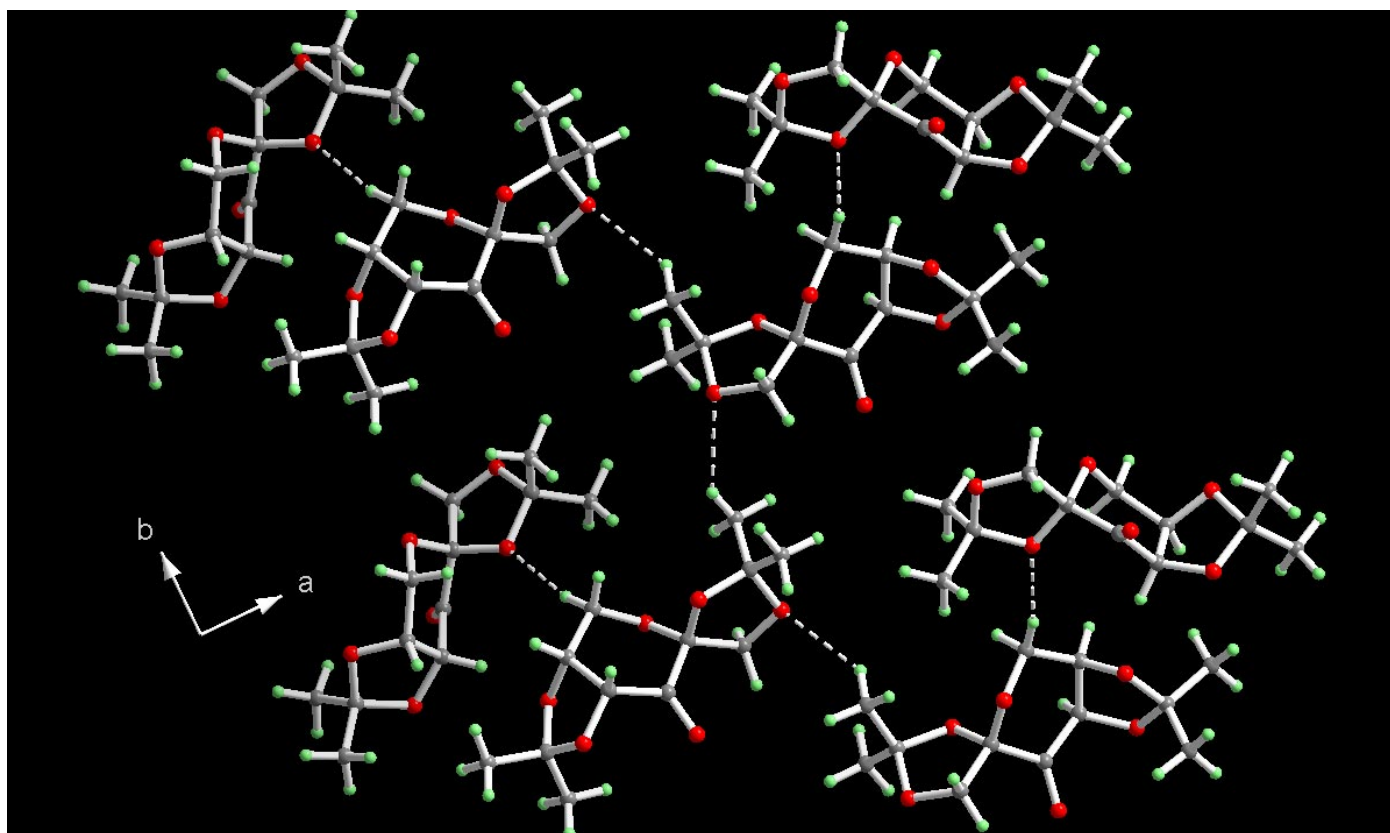


Figure 2
Hydrogen-bond scheme projected along the *c* axis. Molecules A form a zigzag infinite chain along the *b* axis with molecules B as side chains.

crystal structure of the title compound. Molecules *A* and *B* are connected through hydrogen bonds in such a way that molecules *A* form an infinite zigzag chain along the *b* axis with side chains consisting of molecules *B* (Fig. 2). This could be also an explanation of the slightly larger thermal motion of atoms of molecule *B* in comparison with those of molecule *A*.

Experimental

Compound (I) was prepared in two steps from D-fructose according to the procedure of Wang *et al.* (1997).

Crystal data

| | |
|---------------------------------|---|
| $C_{12}H_{18}O_6$ | $D_x = 1.322 \text{ Mg m}^{-3}$ |
| $M_r = 258.27$ | Mo $K\alpha$ radiation |
| Monoclinic, $P2_1$ | Cell parameters from 6790 reflections |
| $a = 10.9909 (2) \text{ \AA}$ | $\theta = 1.8\text{--}25.5^\circ$ |
| $b = 10.5488 (2) \text{ \AA}$ | $\mu = 0.11 \text{ mm}^{-1}$ |
| $c = 11.3876 (2) \text{ \AA}$ | $T = 297 (2) \text{ K}$ |
| $\beta = 100.526 (1)^\circ$ | Needle, colorless |
| $V = 1298.07 (4) \text{ \AA}^3$ | $0.55 \times 0.16 \times 0.15 \text{ mm}$ |
| $Z = 4$ | |

Data collection

| | |
|---|--|
| Siemens SMART CCD diffractometer | 3804 reflections with $I > 2\sigma(I)$ |
| ω scans, 0.30° frames, 30 s each | $R_{\text{int}} = 0.024$ |
| Absorption correction: multi-scan (Blessing, 1995) | $\theta_{\text{max}} = 25.5^\circ$ |
| $T_{\text{min}} = 0.944$, $T_{\text{max}} = 0.984$ | $h = -13 \rightarrow 13$ |
| 13 132 measured reflections | $k = -12 \rightarrow 12$ |
| 4781 independent reflections | $l = -13 \rightarrow 13$ |
| | Intensity decay: none |

Refinement

| | |
|--|--|
| Refinement on F^2 | $w = 1/[\sigma^2(F_o^2) + (0.0502P)^2 + 0.0759P]$ |
| $R[F^2 > 2\sigma(F^2)] = 0.036$ | where $P = (F_o^2 + 2F_c^2)/3$ |
| $wR(F^2) = 0.091$ | $(\Delta/\sigma)_{\text{max}} = 0.009$ |
| $S = 1.02$ | $\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$ |
| 4781 reflections | $\Delta\rho_{\text{min}} = -0.12 \text{ e \AA}^{-3}$ |
| 370 parameters | Extinction correction: <i>SHELXL97</i> |
| H atoms treated by a mixture of independent and constrained refinement | Extinction coefficient: 0.0159 (17) |
| | Absolute structure: (Flack, 1983) |
| | Flack parameter = $-0.2 (8)$, 2221 Friedel pairs |

Table 1

Selected torsion angles ($^\circ$).

| | | | |
|------------------|-------------|------------------|-----------|
| C4A—O4A—C10A—O5A | $-3.2 (2)$ | O2B—C7B—O1B—C1B | $0.8 (3)$ |
| C4B—O4B—C10B—O5B | $-2.9 (2)$ | H5A—C5A—C6A—H6A1 | -70 |
| O1A—C1A—C2A—O2A | $3.4 (2)$ | H5B—C5B—C6B—H6B1 | -71 |
| O1B—C1B—C2B—O2B | $-17.3 (3)$ | H5A—C5A—C6A—H6A2 | 48 |
| O2A—C7A—O1A—C1A | $32.5 (2)$ | H5B—C5B—C6B—H6B2 | 46 |

Table 2

Hydrogen-bonding geometry (\AA , $^\circ$).

| $D\text{---}H\cdots A$ | $D\text{---}H$ | $H\cdots A$ | $D\cdots A$ | $D\text{---}H\cdots A$ |
|-------------------------------------|----------------|-------------|-------------|------------------------|
| C6A—H6A1 \cdots O2B ⁱ | 0.97 | 2.42 | 3.382 (3) | 169 |
| C8A—H8A2 \cdots O1A ⁱⁱ | 0.96 | 2.57 | 3.384 (4) | 142 |

Symmetry codes: (i) $x - 1, y, z$; (ii) $1 - x, \frac{1}{2} + y, 2 - z$.

Data were collected using a Siemens SMART CCD diffractometer at room temperature. A full sphere of reciprocal space was scanned by 0.3° steps in ω with a crystal-to-detector distance of 3.97 cm and 30 s per frame. A preliminary orientation matrix was obtained from the first 100 frames using *SMART* (Siemens, 1995). The collected frames were integrated using the preliminary orientation matrix which was updated every 100 frames. Final cell parameters were obtained by refinement on the position of 6826 reflections with $I > 10\sigma(I)$ after integration of all the frames data using *SAINTE* (Siemens, 1995). The data were empirically corrected for absorption and other effects using *SADABS* (Sheldrick, 1996) based on the method of Blessing (1995). H atoms were constrained to the ideal geometry using an appropriate riding model. For methyl groups, the C—H distances (0.93 \AA) and C—C—H angles (109.5°) were kept fixed, while the torsion angles were allowed to refine with the starting position based on the threefold averaged circular Fourier synthesis.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SAINTE* (Siemens, 1995); data reduction: *SAINTE* and *XPREP* (Siemens, 1995); program(s) used to solve structure: *XFPA98* (Pavelčík, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2000); software used to prepare material for publication: *PLATON* (Spek, 2001).

References

- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.
- Brandenburg, K. (2000). *DIAMOND*. Version 2.1d. Crystal Impact GbR, Bonn, Germany.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Hervé du Penhoat, P. C. M. & Perlin, A. S. (1979). *Carbohydr. Res.* **71**, 135–148.
- Košíš, M., Steiner, B., Langer, V., Gyepesová, D., Ďurík, M. (2000). *Carbohydr. Res.* **328**, 115–126.
- Košíš, M., Steiner, B., Gajdoš, J., Langer, V., Gyepesová, D., Smrčok, Ľ., Ďurík, M. (2000). *Molecules*, **5**, 219–226.
- McDonald, E. J. (1967). *Carbohydr. Res.* **5**, 106–108.
- Pavelčík, F. (1999). *J. Appl. Cryst.* **32**, 839–840.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Siemens (1995). *SMART*, *SAINTE* and *XPREP*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2001). *PLATON*. Utrecht University, The Netherlands.
- Steiner, B., Košíš, M., Langer, V., Gyepesová, D. & Smrčok, Ľ. (1998). *Carbohydr. Res.* **311**, 1–9.
- Wang, Z. X., Tu, Y., Zhang, J. R. & Shi, Y. (1997). *J. Am. Chem. Soc.* **119**, 11224–11235.